

DII-003-011404

Seat No. ____

M. Sc. (Industrial Chemistry) (Sem. IV) (CBCS) Examination

May / June - 2015

IC - 404 : Advanced Organic Chemistry

Faculty Code: 003 Subject Code: 011404

Time : $2\frac{1}{2}$ Hours]

[Total Marks: 70

Instructions: 1) All Questions are compulsory & carries 14 marks.

Q.1] Answer any Seven out of the following ten questions:

14

- 1. Define stereochemistry.
- 2. What is optical activity? Define dextrorotatory and levorotatory.
- 3. Describe the term Atropisomers and racemic mixture.
- 4. Give two applications of Reformatsky reaction.
- 5. Draw the staggered and eclipsed conformation of cyclohexane.
- 6. Give two applications of Sonogashira cross coupling reaction.
- 7. Discuss optical isomerism in biphenyl compounds.
- 8. Define green chemistry and sustainability.
- 9. Give Principal and mechanism of Sharpless epoxidation.
- 10. Explain Cis and Trans isomerism with example.

Q2] Answer any Two from the following three questions:

14

- 1. Discuss the stereochemistry of Allenes.
- 2. Explain principle of green chemistry.
- 3. Give fisher projection of following compound and assign configurations.

Q3] Answer the following Two questions:

14

- 1. Explain the conformation of 1:3 dimethyl cyclohexane.
- 2. Discuss Stille cross coupling reaction.

OR

Q3]	Answer the following Two questions:	14
1	Define the term storegissmer and give electification with examples	

- 1. Define the term stereoisomer and give classification with examples.
- 2. Give three applications of $NaCNBH_3$ and $Tl(NO_3)_3$.

Q4] Answer any Two from the following three questions: 14

- 1. Define Asymmetric induction and discuss Prelog's rule.
- 2. Give preparation and applications of lithium organocuprates reagent.
- 3. Give Principle, Mechanism and application of Strecker reaction.

Q5] Answer any Two from the following four questions: 14

- 1. Explain: (i) Principle and advantages of microwave assisted organic reaction (ii) Miyaura coupling reaction with example.
- 2. Discuss in detail DIBAL.
- 3. Describe D and L nomenclature giving examples.
- 4. Give Heck reaction.